T(6th Sm.)-Chemistry-H/(DSE-A-3)/CBCS

2021

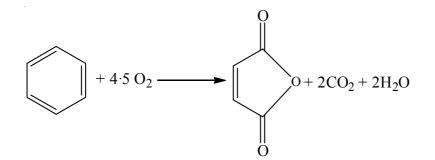
CHEMISTRY — HONOURS Paper: DSE-A-3 (Green Chemistry and Chemistry of Natural Products)

Full Marks : 50

The figures in the margin indicate full marks. Candidates are required to give their answers in their own words as far as practicable.

Answer question no. 1 (compulsory) and any eight questions from the rest (question nos. 2 to 13).

- 1. Answer *any ten* questions:
 - (a) What is CCS in green chemistry?
 - (b) What is the green chemistry principle regarding waste management?
 - (c) What do you mean by the term PEG-400?
 - (d) Write down one advantage of using biomimetic synthesis.
 - (e) What is the source of microwave irradiation?
 - (f) Give an example of a "susceptor".
 - (g) Why is an ionic liquid called green solvent?
 - (h) Define the term percent atom economy.
 - (i) Which type of "cavitation" is mainly responsible for a chemical reaction?
 - (j) What is isoprene rule?
 - (k) Mention one medicinal use of reserpine.
 - (1) Give example of one green reagent which can be used in Friedel-Crafts reaction instead of AlCl₃.
- **2.** (a) (i) Calculate the % atom economy of the following reaction:



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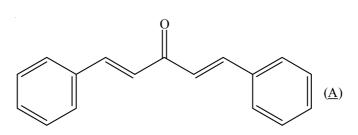
	(ii) Give an example of a reaction having 100% atom economy.	3
	(b) What is the basic difference between "% Atom Economy" and "E-Factor"?	2
3.	(a) Decomposition of H_2S over γ -Al ₂ O ₃ or MoS ₂ - γ -Al ₂ O ₃ by MW irradiation is an efficient pr to formation of hot spots. How can you justify it?	ocess due 3
	(b) What is the difference between a "susceptor" and a "catalyst"?	2
4.	(a) What is the main principle of "combinatorial chemistry"?	3
	(b) Write two advantages of solventless reactions.	2
5.	(a) Suggest a convenient synthesis using aldol condensation for the following compound	(A). (No

(2)

- (a) Suggest a convenient synthesis using aldol condensation for the following compound (<u>A</u>). (No mechanism required.)
 - (b) Suggest one alternative greener approach of aldol condensation for the synthesis of compound (\underline{A}).

2

2



- 6. (a) Give one example of decarboxylation reaction using microwave irradiation (MWI). Write the green context of the reaction.
 - (b) Can mircrowave irradiation (MWI) affect the structure of an organic molecule? Explain. 2
- (a) Reaction between cyclopentadiene and methyl vinyl ketone shows decrease in activation energy from changing the reaction medium from gaseous phase to aqueous solution by 2.8 kcal/mol and reaction between cyclopentadiene and isoprene shows decrease in activation energy by 4.5 kcal/mol. What factors are responsible for this observation?
 - (b) Give an example of Claisen rearrangement in water. 2
- **8.** (a) How do the structures of cations and anions control the property of ionic liquids (IL)? 3
 - (b) What type of gas molecules are prone to sonoluminescence and why?
- 9. (a) Write down the disadvantages of the conventional method of Beckmann rearrangement. Elaborate with one example about the green approach of the reaction.
 - (b) Mention two disadvantages of the conventional method to carry out Fries rearrangement.
- 10. (a) What are the new areas on which future trends of green chemistry depend?
 3
 - (b) Write two disadvantages of commonly used oxidising agents. Name one green oxidising agent. 2

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- 11. (a) How can adipic acid be synthesised following green procedure? Write down the steps involved.3
 - (b) Write down Diels-Alder reaction between anthracene and dimethyl fumarate under microwave irradiation. Mention how the reaction is improved under MWI compared to traditional Diels-Alder reaction. 2

12.	(a)	Write down the structures of two geometrical isomers of citral. How will you synthesise citral start	ing
		from 6-methyl-hept-5-ene-2-one?	3
	(b)	What is multifunctional reagent? Give an example.	2
13.	(a)	How can you detect the functional nature of oxygen present in an alkaloid chemically?	3
	(b)	Define alkaloids.	2

(3)